

JACK SHREVE PUBLIC COUNSEL

STATE OF FLORIDA OFFICE OF THE PUBLIC COUNSEL

c/o The Florida Legislature 111 West Madison St. Room 812 Tallahassee, Florida 32399-1400 850-488-9330 August 30, 1999

ORIGINAL

CIDEVED-17930

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REPORTING

Blanca S. Bayo, Director Division of Records and Reporting Florida Public Service Commission 2540 Shumard Oak Blvd. Tallahassee, FL 32399-0850

Re: Case No. 960545-WS

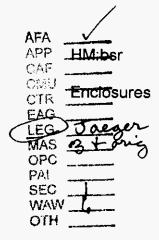
Dear Ms. Bayo:

Enclosed for filing in the above-referenced docket are original and 15 copies of the Direct Testimony of Ted L. Biddy, P.E./P.L.S. on Behalf of the Citizens of the State of Florida.

Please indicate the time and date of receipt on the enclosed duplicate of this letter and return it to our office.

Sincerely,

ffarold McLean Associate Public Counsel



RECEIVED & **OF RECORDS**

DOCUMENT NUMBER-DATE

BEFORE THE FLORIDA PUBLIC SERVICE COMMISSIONAL

In Re Investigation of Utility rates of Aloha Utilities, Inc. In Pasco County, Florida.

DOCKET NO. 960545-WS

FILED: August 30, 1999

DIRECT TESTIMONY

OF

TED L. BIDDY, P.E./P.L.S.

On Behalf of the Citizens of The State of Florida

Docket No. 960545-WS

Jack Shreve Public Counsel

Office of Public Counsel c/o The Florida Legislature 111 West Madison Street Room 812 Tallahassee, FL 32399-1400

(850) 488-9330

Attorney for the Citizens of the State of Florida

DOCUMENT NUMBER-DATE

FPSC-RECORDS/REPORTING

a Table

1		DIRECT TESTIMONY OF TED L BIDDY, P.E./P.L.S.
2		ON BEHALF OF THE CITIZENS OF FLORIDA
3		
4		BEFORE THE
5		FLORIDA PUBLIC SERVICE COMMISSION
6		DOCKET NO. 960545-WS
7		
8	Q.	What is your name, and business address?
9	А.	My name is Ted L. Biddy. My business is 2308 Clara Kee
10		Boulevard, Tallahassee, Florida 32303.
11	Q.	By whom are you employed and what is your position?
12	A.	I am self-employed as a professional engineer and land surveyor.
13	Q.	What is you educational background and work experience?
14	А.	I graduated from the Georgia Institute of Technology with a B.S.
15		degree in Civil Engineering in 1963. I am a registered professional
16		engineer and land surveyor in Florida, Georgia and Mississippi and
17		several other states. I was the vice-president of Baskerville-
18		Donovan, Inc. (BDI) and the regional manager of the Tallahassee
19		Office from April 1, 1991 until February, 1998. I left the
20		employment of BDI on September 30, 1998. Before joining BDI in
21		1991, I had operated my own civil engineering firm for 21 years.
22		My areas of expertise include civil engineering, structural
23		engineering, sanitary engineering, soils and foundation engineering
24		and precise surveying. During my career, I have designed and
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1		supervised the master planning, design and construction of			
2		thousands of residential, commercial and industrial properties. My			
3		work has included water and wastewater design; structural design;			
4		land surveys; and environmental permitting.			
5		I have served as principal and chief designer for numerous utility			
6		projects. Among my major water and wastewater facilities designs			
7		have been a 2,000 acre development in Lake County, FL; a 1,200			
8		acre development in Ocean Springs, MS; a 4 mile water distribution			
9		system for Talquin Electric Cooperative, Inc. and a 320 lot			
10		subdivision in Leon County, FL.			
11	Q.	What are your professional affiliation?			
12					
13	 system for Talquin Electric Cooperative, Inc. and a 320 lot subdivision in Leon County, FL. Q. What are your professional affiliation? A. I am a member of the Florida Engineering Society, National Society of Professional Engineers, Florida Institute of Consulting Engineers, American Consulting Engineers Council, American College of Forensic Examiners and the Florida Society of Professional Land Surveyors. Q. Have you previously testified before the Florida Public Service Commission (FPSC)? 				
14	Engineers,	American Consulting Engineers Council, American College of			
15		Forensic Examiners and the Florida			
16	Society of]	Professional Land Surveyors.			
17	Q.	Have you previously testified before the Florida Public Service			
18		Commission (FPSC)?			
19	А.	Yes. I have testified before the PSC for Docket Nos. 940109-WU,			
20		950495-WS, 9503870-SU, 951056-SU, 960329-WS and the			
21		remand cases Docket Nos. 950387-SU and 971065-SU on various			
22		engineering issues and used and useful analyses.			
23	Q.	Have you previously testified before a state or federal court as an			
24		engineering expert witness?			
25		Yes, I have had numerous court appearances as an expert witness			
		2			

1		for cases involving roadways, utilities, drainage, stormwater, water
2		and wastewater facilities designs.
3	Q.	What is the purpose of your testimony?
4	А.	The purpose is to present testimony in connection with a water
5		quality investigation which I have concluded of the Aloha Utilities
6		Seven Springs Water System specifically in connection with "black
7		water" complaints by the customers of Aloha Utilities
8	Q.	What documents have you reviewed in preparation for your
9		testimony?
10	А.	I have reviewed the pre-filed direct testimony of Aloha's President
11		Stephen G. Watford and the pre-filed direct testimony of Aloha's's
12		Consulting Engineer David W. Porter, P.E., along with all of their
13		exhibits which included an extensive study by Mr. Porter entitled
14		"Water Facilities Upgrade Study Report." I also reviewed the
15		"Notice of Proposed Agency Action" of the Public Service
16		Commission dated January 7, 1999
17	Q.	During your review of the "Water Facilities Upgrade Study
18		Report", did you form an opinion as to whether this document
19		adequately addressed the Commission's March 12, 1997 order to
20		Aloha to prepare a report that evaluated the costs and efficiencies of
21		several treatment options for the removal of hydrogen sulfide from
22		its source water?
23	А.	Yes, I formed the opinion that the report did not adequately address
24		the Commission's order in that the report did not attempt to isolate
25		the problem area(s) and then study ways to upgrade the water
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1	quality at the problem area(s) but the study included extensive new
2	water treatment, storage, and pumping facilities for all nine existing
3	well sites and even added additional well sites. Indeed, the
4	recommended system, known as Alternative 2 - Centralized Water
5	Facilities, includes new and expanded facilities for this utility's
6	needs through the year 2015 and beyond at a cost of 9.5 million
7	dollars. This broad brush approach would obviously be good for
8	the utility but in no way solves the problem in a cost effective
9	manner. I believe that the study should have concentrated a
10	study into the cause and cure of the water quality problems at the
11	southwest portion of the service area served by well nos. 8 and 9
12	where most of the water quality complaints have come from.
13	Indeed, the Chelsea, Wyndtree and Wyndgate Subdivisions and
14	surrounding areas are the locations of most of the complaints of
15	"black water" and all of these areas are served by wells nos. 8 and
16	9. During my investigation, I asked several of the customers in
17	these areas when they first started experiencing the "black water"
18	problem. Without exception, each consumer stated that is had only
19	been 4 to 5 years ago and they then volunteered that is was at the
20	time that the new wells nos. 8 and 9 were added to the system.
21	Therefore, the water source from wells nos. 8 and 9
22	is highly suspect as being involved with the water quality problems
23	in this area. At the very least, a detailed study of these wells should
24	be performed. If a single packed tower aeration unit at these wells
25	could solve the water quality problem, the cost would only be a

fraction of the estimates which the utility's study concluded for the entire system.

Q. What independent studies did you perform during your investigation of the water quality of Aloha Utilities Seven Springs water system?

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A. I employed an independent testing laboratory, Savannah
 Laboratories & Environmental Services, Inc. to accompany me to
 the Aloha service area to collect samples at five of the eight Aloha
 wells for testing. It was necessary for the Office of Public
 Counsel's (OPC) attorney to arrange for access to the well sites
 with the attorney for Aloha Utilities who wanted to be present
 during the sampling along with Aloha officials and their testing lab.

13 With these arrangements made, I traveled to the Aloha 14 service area along with OPC's attorney and the lab technician from 15 Savannah Laboratories. We were met at the site by Alohas' 16 attorney, Corporate President and their system superintendent along 17 with a technician from their testing lab. Additionally, Attorney 18 Ralph Jaeger and Engineer Robert Crouch were present 19 representing the Public Service Commission. Together we all 20 traveled to well sites 1, 3, 6, 8 & 9 and obtained raw and finished 21 water samples for testing. Photographs of each of these well sites 22 were taken. (Attached hereto as Exhibit TLB-1). 23 Q. What parameters did the Savannah Laboratories test for and what 24 was the results of the testing of the raw and finished water from 25 these wells?

1	А.	The test results (Attached hereto as Exhibit TLB-2) are very
2		puzzling. The Lab tested for Copper, Hardness, Sulfate, Color,
3		Odor, pH and Sulfides in both the raw and finished water. The test
4		results are remarkable for their lack of detection of sulfides and
5		sulfates in both the raw and finished water samples. All samples
6		tested for each of the five wells found sulfides below 0.10 mg/l
7		which is the detection limit. Similarly, the tests for sulfates at all
8		wells were below 5 mg/l which is the detection limit for this
9		parameter. This is indeed a strange finding when the utility has
10		previously admitted that their raw well water contains modest
11		amounts of hydrogen sulfide and the finished water has these
12		sulfides converted to sulfates. The test results show a moderately
13		hard water with hardness ranging from 180 mg/l to 220 mg/l. The
14		test results for copper were also below the detection limit of 0.02
15		mg/l for both raw and finished water for all wells except for a
16		copper level of 0.046 mg/l in the finished water from well no. 9.
17		Test for pH and color also showed these parameters to be within the
18		acceptable range.
19		These test results would lead one to think that the water
20		from these five Aloha Wells was almost pristine and of a very high
21		quality. However, the tests for odor from the raw and finished
22		water of all of the wells except for well no. 6 have Threshold Order
23		Numbers in excess of the Florida DEP Secondary Drinking Water
24		Standard of 3. The tests for the finished water from well nos. 1, 3
25		& 9 showed Threshold Odor Numbers of 16 which is more than 5

1		times the FDEP standard for drinking water.
2	Q.	What do you conclude from these test results?
3	А.	I was at first extremely puzzled by the seemingly high quality of the
4		well water tested in light of the history of this utility with water
5		quality problems at the customer residences consisting of odor,
6		taste and discolored water. I then investigated the very high odor
7		test results from well nos. 1, 3 & 9. The Laboratory informed me
8		that the odor was a very strong chlorine odor.
9		It then became clear to me that the utility or someone had
10		heavily dosed the raw and finished water of the wells with chlorine
11		on the morning of August 4, 1999 prior to our arrival at the site for
12		sampling. The heavy chlorine dosing would have lowered the
13		sulfides and sulfates below detection limits and therefore the testing
14		was rendered worthless. Certainly the utility cannot operate the
15		water system with such high chlorine dosing. In fact, the utility had
16		to lower the chlorine dosage rate in September, 1995 in response to
17		EPA requirements for lowering of trihalomethanes which are
18		disinfection by-products which are formed when natural organics in
19		water react with chlorine. The trihalomethanes are considered by
20		the EPA to be primary contaminants and possibly carcinogenic. It
21		is certainly interesting to note that the discolored water complaints
22		began to increase shortly after the chlorine dosage was lowered
23		which is of course also the time when wells nos. 8 and 9 were
24		added to the system in December, 1995.
25	Q.	Can you prove that Aloha Utilities intentially tried to rig the test

1		results by adding a heavy dosage of chlorine to the raw and finished
2		well water before you and Savannah Laboratories arrived at the
3		area?
4	А.	No, I cannot prove that the utility added the heavy doses of chlorine
5		to the raw and finished well water. I do have strong suspicions
6		about the matter based on the results of the odor tests of the raw and
7		finished water. Obviously, someone added the chlorine.
8	Q.	Mr. Biddy, can you say to a reasonably scientific certainty that both
9		the raw and finished well water was subjected to extraordinary
10		chlorination, such that the testing you had performed for sulfides
11		was affected?
12		
13	А.	Yes, I can. That is my professional opinion.
14	Q.	What specifics lead you to this conclusion?
15	A.	Three considerations: first, that the Utility itself reported to this
16		commission that its water contained sulfides and sulfates, whereas
17		the samples taken and analyzed did not. Second, it is my
18		experience over many years of working in environmental
19		engineering in the costal areas of our state that water from these
20		areas invariably contains significant, measurable sulfides and
21		sulfates, whereas the samples taken and analyzed did not. Lastly,
22		that the water samples taken retained significant evidence of
23		chlorination, long after chlorine residue should have dissipated
24		entirely.
25	Q.	Have you excluded every other reasonable hypothesis which would

1		explain the characteristics of the water you had tested?								
2	А.	I believe that I have. It is my professional opinion that the water								
3	which was tested had been chlorinated to an extent atypical of that									
4		normally provided by Aloha Utilities.								
5										
6	Q.	What other actions did you take in connection with your								
7	investigatio	on of the water quality of the Aloha Utilities Seven Springs								
8	Water Syst	tem?								
9	А.	On the morning following the sampling of the Aloha Wells which								
10		was August 5, 1999, I took the same Savannah Laboratories								
11		technician to six different residences in the Southwest portion of								
12		Aloha's service area that is served by well nos. 8 & 9 for purposes								
13		of taking water samples for testing from inside and outside the								
14		homes. I interviewed each homeowner and obtained three water								
15		samples from each house. I also took photographs at the residences								
16		which are attached hereto as Exhibit TLB-3. The water samples								
17		were taken from both a hot water and cold water faucet in the house								
18		and an additional cold water sample was taken from a yard hose bib								
19		between the house and the meter. The Savannah Laboratories								
20		technician then carried the samples to the lab for testing.								
21	Q.	What parameter were tested for at these homes and what did the								
22		testing show?								
23	А.	The same parameters were first tested for as with the water at the								
24		wells with Copper, Hardness, Sulfates, Color, Odor, pH and								
25		Sulfides begin obtained for all three samples at each home. The								
		9								

results for sulfides and sulfates showed the same low levels as with the well samples since this water had been super-chlorinated less that twenty-four hours before. Color and pH were also found in the same acceptable range as with the well tests. Odor was now undetectable since the excess chlorine and reacted with all water in the distribution system in that area and was essentially used up. Hardness in the water from outside hose bibs was hard with values between 170 mg/l and 240 mg/l. In several homes with water softening units, the water inside the homes was very soft. The big variable in the testing was at the level of copper. The test values ranged from 0.02 mg/l at the Straighter residence on Byrnwood Dr. to an astonishing high of 16.0 mg/l at the cold water faucet in the Coogan residence on Davenport Dr. The hot water sample in the Coogan residence showed a cooper level of 10.0 mg/l. These high copper contents are very much more than the FDEP action required level of 1.3 mg/l and the secondary drinking water standard of 1.0 mg/l. Further testing was then ordered from Savannah Laboratories consisting of qualitative anion screen to determine what anions might be present in the black water residue in the Coogan residence. The lab found chlorides at a level of 30 mg/l and sulfates at 20 mg/l and <u>no sulfides</u>. Therefore, the black water is not being caused by copper sulfide as has been the claim of Aloha Utilities in the past. Further testing was also ordered on the finished water of well no. 9 which had a strong chlorine odor. The lab found 1.4 mg/l of free chlorine in the sample even though

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testing was performed three weeks after the sample was taken. This chlorine level would have been many times higher if tested on the day the sample was taken. See Exhibits TLB-4 & TLB-5 for lab tests.

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What can you conclude from the water testing at the residences? Q. It is difficult to tell anything about the sulfate and sulfide A. concentrations since the well water had, in my opinion, been excessively chlorinated the day before the testing. What can be concluded from the limited testing program at the residences is that all the water faucets tested showed some amount of copper and the Coogan residence had very excessive amounts in both the cold and hot water taps. These copper levels of 10 to 16 mg/l are greatly in excess of the FDEP Secondary Drinking Water Standard for copper of 1.0 mg/l. The black residue in the Coogan water was screened for anions and no sulfides were found. Therefore, at least at the Coogan residence, the very black water is not due to copper sulfide but from some other unknown chemical combined with cooper.. What other investigations did you perform in connection with this Q. matter?

I investigated methods available for removing the sulfides and 20 A. sulfates from the well water and the cost of these facilities. I found that pressure filters would be by far the most cost effective method 22 for sulfide and sulfate removal. The pressure filters are installed on line with the existing pressurized system and therefore do not require an additional storage tank or high service pumping as

1	1	packed tower aeration require. The pressure filters are said to
2		remove all sulfides and sulfates and the cost for complete
3		installation at a 500 gpm well would be about #225,000 to
4		\$250,000. This cost is only about 25 percent of the cost of the
5		packed tower installation cost at a 500 gpm well as contained in the
6		"Water Facilities Upgrade Study Report" as prepared by Aloha
7		Utilities and attached to Mr. Porter's Testimony as Exhibit DWP-1.
8	Q.	What would be your recommendation to the Public Service
9		Commission in this matter?
10	А.	Well, first I would tell the Commission that the source of the
11		problem has not been determined yet by me or by the utility. This
12		is clear from the analysis of the black water residue at the Coogan
13		residence which does not contain any sulfides which have
14		previously been blamed by the utility as being combined with
15		copper at the residences to cause the black water problem.
16		Therefore, I would recommend to the Commission that the Utility
17		again be ordered to study the problem in a comprehensive manner
18		by concentrating an investigation in the areas of black water
19		complaints including the wells which serve these areas and
20		determine for certain what is causing the water quality problems
21		and that the utility develop a cost effective solution to these
22		problems. I would recommend that the study further include a
23		comparison with other nearby systems such as the Pasco County
24		system particularly in regards to this system's customers who have
25		home water softening units. And finally, I would request that the

1		Commission order the Utility to make their well sites accessible to
2		Office of Public Counsel's experts and that the Utility do nothing to
3		frustrate any testing which these experts may make.
4	Q.	Do you have anything else to add to your testimony?
5	А.	No, not at this time.
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CERTIFICATE OF SERVICE DOCKET NO. 960545-WS

I HEREBY CERTIFY that a copy of the foregoing has been furnished by

U.S. Mail or hand-delivery to the following parties on this 30th day of August,

1999.

Harold McLean

Ralph Jaegar Division of Legal Services Fla. Public Service Commission 2540 Shumard Oak Blvd. Tallahassee, FL 32399-0850

Representative Mike Fasano 8217 Massachusetts Avenue New Port Richey, FL 34653 Marshall Deterding, Esq. Rose, Sundstrom & Bentley 2548 Blairstone Pines Drive Tallahassee, FL 32301

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EXHIBIT TLB-1: PHOTOGRAPHS OF WELL SITES

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WELL NO. 1 SAMPLING





WELL NO. 6





WELL NO. 3





WELL NO. 8





YARD VIEW OF WELL NO. 9



WELL NO. 9

EXHIBIT TLB-2: WELL TEST RESULTS

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SAVANNAH LABORATORIES & ENVIRONMENTAL SERVICES, INC.

Mr. Ted Biddy

 52379-1
 MW-1 (RAW)

 52379-2
 MW-1 (FINISH)

 52379-3
 MW-6 (RAW)

52379-4 MW-6 (FINISH) 52379-5 MW-3 (RAW)

LOG NO

PARAMETER

2308 Clara Kee Blvd Tallahassee, FL 32303

6712 Benjamin Road • Suite 100 • Tampa, FL 33634 • (813) 885-7427 • Fax (813) 885-7049 • www.savlabs.com

LOG NO: B9-52379 Received: 04 AUG 99 Reported: 23 AUG 99 Florida Legislature-Office Of Public Counsel Project: Aloha Utilities Sampled By: Savannah Laboratories Code: 132790823 REPORT OF RESULTS Page 1 DATE/ TIME SAMPLED SAMPLE DESCRIPTION , LIQUID SAMPLES _____ 08-04-99/09:50 08-04-99/10:03 08-04-99/10:33 08-04-99/10:36 08-04-99/11:07 _____ 52379-1 52379-2 52379-3 52379-4 52379-5

Copper (200.7)					
Copper, mg/l	<0.020	<0.020	<0.020	<0.020	<0.020
Prep Date	08.10.99	08.10.99	08.10.99	08.10.99	08.10.99
Analysis Date	08.17.99	08.17.99	08.17.99	08.17,99	08.17.99
Hardness as CaCO3, mg/l	220	220	180	180	190
Sulfate as SO4, mg/l	<5.0	<5.0	<5.0	<5.0	<5.0
Analysis Date	08.09.99	08.09.99	08.09.99	08.09.99	08.09.99
Color, PCU	25	10	20	10	15
Analysis Date	08.05.99	08.05.99	08.05.99	08.05.99	08.05.99
Odor, TON	4	16	2	2	4
Analysis Date	08.04.99	08.04.99	08.04.99	08.04.99	08.04.99
pH, units	7.2	7.2	7.6	. 7.2	7.6
Analysis Date		08.04.99			
Sulfide (376.2), mg/l	<0.10	<0.10	<0.10	<0.10	<0.10
Prep Date	08.10.99		08.10.99		
Analysis Date	08.10.99			08.10.99	



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LOG NO: B9-52379 Received: 04 AUG 99 Reported: 23 AUG 99 Mr. Ted Biddy Florida Legislature-Office Of Public Counsel 2308 Clara Kee Blvd Tallahassee, FL 32303 Project: Aloha Utilities Sampled By: Savannah Laboratories Code: 132790823 REPORT OF RESULTS Page 2 DATE/ LOG NO SAMPLE DESCRIPTION , LIQUID SAMPLES TIME SAMPLED 52379-6 MW-3 (FINISH) 08-04-99/11:12 52379-7 MW-8 (RAW) 08-04-99/11:40 52379-8 MW-8 (FINISH) 52379-9 MW-9 (RAW) 08-04-99/11:43 08-04-99/12:08 52379-10 MW-9 (FINISH) 08-04-99/12:12 PARAMETER 52379-6 52379-7 52379-8 52379-9 52379-10 Copper (200.7) Copper, mg/l <0.020 <0.020 <0.020 <0.020 0.046 08.10.99 08.10.99 08.10.99 08.10.99 08.10.99 Prep Date Analysis Date 08.17.99 08.17.99 08.17.99 08.17.99 08.13.99 Hardness as CaCO3, mg/l 210 190 210 210 210 Sulfate as SO4, mg/l <5.0 <5.0 <5.0 <5.0 <5.0 Analysis Date 08.09.99 08.09.99 08.09.99 08.09.99 08.09.99 Color, PCU 10 20 - 5 20 - 5 Analysis Date 08.05.99 08.05.99 08.05.99 08.05.99 08.05.99 Odor, TON 4 4 16 4 16 Analysis Date 08.04.99 08.04.99 08.04.99 08.04.99 08.04.99 pH, units 7.6 7.7 7.2 7.2 7.1 08.04.99 08.04.99 Analysis Date 08.04.99 08.04.99 08.04.99 Sulfide (376.2), mg/1 <0.10 <0.10 <0.10 <0.10 <0.10 Prep Date 08.10.99 08.10.99 08.10.99 08.10.99 08.10.99 Analysis Date 08.10.99 08.10.99 08.10.99 08.10.99 08.10.99

SAVANNAH LABORATORIES & ENVIRONMENTAL SERVICES, INC.

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LOG NO: B9-52379 Received: 04 AUG 99 Reported: 23 AUG 99 Mr. Ted Biddy Florida Legislature-Office Of Public Counsel 2308 Clara Kee Blvd Tallahassee, FL 32303 Project: Aloha Utilities Sampled By: Savannah Laboratories Code: 133890823 REPORT OF RESULTS Page 3 DATE/ LOG NO SAMPLE DESCRIPTION , QC REPORT FOR LIQUID SAMPLES TIME SAMPLED 52379-11 Method Blank 52379-12 Accuracy (%Rec) 52379-13 Precision (%RPD) ______ PARAMETER 52379-11 52379-12 52379-13 _____ Copper (200.7) Copper, mg/l <0.020 108 % 2.9 % 08.10.99 08.10.99 Prep Date _ _ _ Analysis Date 08.10.99 08.10.99 _ _ _ Hardness as CaCO3, mg/l <3.3 110 % 1.2 % Sulfate as SO4, mg/l <5.0 91 % 8.9 % Analysis Date 08.09.99 08.09.99 ---Color, PCU 100 % <5 0 % Analysis Date 08.05.99 08.05.99 - - -Odor, TON <10 _ _ _ - - -Analysis Date 08.04.99 - - -- - pH, units 102 % 6.1 0 % 08.04.99 08.04.99 Analysis Date _ _ _ Sulfide (376.2), mg/l <0.10 109 % 0.92 % Prep Date 08.10.99 08.10.99 ---Analysis Date 08.10.99 08.10.99 - - -Method: EPA 600/4-79-020 DOH Certification #'s: E84282, 84385, 87279, E87052

Kathy Sheffield, & bject Manager

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EXHIBIT TLB-3: PHOTOGRAPHS AT RESIDENCES WHERE TESTING OCCURRED

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ST. ARNO RESIDENCE AT 6809 WILLETS DR.



40 GALLON HOT WATER HEATER AT ST. ARNO RESIDENCE



WATER SOFTING UNIT AT ST. ARNO RESIDENCE



WATER SAMPLING AT YANNA RESIDENCE AT 7437 CHELTENHAM COURT



WATER SOFTENING UNIT AT YANNA RESIDENCE



STRAUDER RESIDENCE AT 2528 BYRNWOOD DRIVE



HOT WATER HEATER AT STRAUDER RESIDENCE (NOTE TWO SHORT COPPER LINES ON HOT WATER HEATER WHICH ARE ONLY COPPER PIPES IN RESIDENCE)



DAVIS RESIDENCE AT 2727 CYPRESS HOLLOW



HOT WATER HEATER AT DAVIS RESIDENCE



WATER SOFTENING UNIT AT COOGAN RESIDENCE



WATER SOFTENING SALT PELLETS AT COOGAN RESIDENCE



HOT WATER HEATER AT COOGAN RESIDENCE AT 1430 DAVENPORT DRIVE

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EXHIBIT TLB-4: LAB TEST RESULTS FOR WATER SAMPLES AT RESIDENCES

SAVANNAH LABORATORIES & ENVIRONMENTAL SERVICES, INC.

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Reported: 18 AUG							
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	308 Clara Kee Blvd		•				
Ta	allahassee, FL 323	303					
		Project: Aloha Utiliti					
				Sampled By: Savannah Laboratories			
					Code	: 130190824	
		REPORT	OF RESULTS			Page 1	
					DATE/		
JOG NO	SAMPLE DESCRIPTI	ON , LIQUID S.	AMPLES		TIME SAMPLE	D	
2407-1	Cold-6809 Willet	s/Mr St Arn		08-05-99/10:15			
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	Outside-6809 Wil				08-05-99/10		
	Cold-7437 Chelta			08-05-99/11:02			
	Hot-7437 Cheltam	•		08-05-99/11:05			
ARAMETER		52407-1	52407-2	52407-3	52407-4	52407-5	
opper (20							
Copper (20		0.052	0.12	0.047	0.46	0.45	
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ardness a	us CaCO3, mg/l	<3.3	22	210	<3.3	5.7	
	; SO4, mg/l	<5.0					
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Odor, TON		<1	<1	<1	<1	<1	
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pH, units		7.0		7.1	7.1	7.3	
Analysis Date		08.05.99	. 08.05.99	08.05.99	08.05.99	08.05.99	
ulfido (a	7() mm/]	-0.10					
Sulfide (376.2), mg/l					0.10 		
Prep Date							

& ENVIRONMENTAL SERVICES, INC.

6712 Benjamin Road • Suite 100 • Tampa, FL 33634 • (813) 885-7427 • Fax (813) 885-7049 • www.savlabs.com

LOG NO: B9-52407 Received: 05 AUG 99 a far far a stand a sta Mr. Ted Biddy Florida Legislature-Office Of Public Counsel 2308 Clara Kee Blvd Tallahassee, FL 32303 Project: Aloha Utilities . • Sampled By: Savannah Laboratories Code: 162490824 REPORT OF RESULTS Page 2 DATE/ SAMPLE DESCRIPTION , LIQUID SAMPLES TIME SAMPLED LOG NO ______ Outside-7437 Cheltam/Mr. Yanna 08-05-99/11:22 52407-6 52407-7Cold-2727 Cypress Hollow/Mr. Davis52407-8Hot-2727 Cypress Hollow/Mr. Davis52407-9Outside-2727 Cypress Hollow/Mr. Davis 08-05-99/12:28 08-05-99/12:32 08-05-99/12:44 52407-10 Cold-2528 Brynwood/Mr. Stouder 08-05-99/13:08 _____ 52407-6 52407-7 52407-8 52407-9 52407-10 PARAMETER Copper (200.7) 0.32 0.20 0.21 0.037 <0.020 Copper, mg/l 08.12.99 08.12.99 08.12.99 08.12.99 08.12.99 Prep Date Analysis Date 08.16.99 08.16.99 08.16.99 08.16.99 08.16.99 Hardness as CaCO3, mg/l 170 170 210 <3.3 230 Sulfate as SO4, mg/l <5.0 <5.0 7.4 6.0 <5.0 Analysis Date 08.09.99 08.09.99 08.09.99 08.09.99 08.09.99 5 Color, PCU 10 10 10 -5 08.05.99 08.05.99 08.05.99 Analysis Date 08.05.99 08.05.99 Odor, TON <1 <1 <1 <1 <1 08.05.99 08.05.99 08.05.99 08.05.99 08.05.99 Analysis Date pH, units 7.3 7.2 7.2 7.4 7.5 Analysis Date 08.05.99 08.05.99 08.05.99 08.05.99 08.05.99 <0.10 Sulfide (376.2), mg/1 <0.10 <0.10 <0.10 <0.10 Prep Date 08.11.99 08.11.99 08.11.99 08.11.99 08.11.99 Analysis Date 08.11.99 08.11.99 08.11.99 08.11.99 08.11.99



& ENVIRONMENTAL SERVICES, INC.

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LOG NO: B9-52407 Received: 05 AUG 99 Ken and the second s Mr. Ted Biddy Florida Legislature-Office Of Public Counsel 2308 Clara Kee Blvd Tallahassee, FL 32303 Project: Aloha Utilities Sampled By: Savannah Laboratories Code: 162490824 REPORT OF RESULTS Page 3 DATE/ LOG NO SAMPLE DESCRIPTION , LIQUID SAMPLES TIME SAMPLED 52407-14 Hot-2528 Brynwood/Mr. Stouder 08-05-99/13:12 52407-15 Outside-2528 Brynwood/Mr. Stouder 08-05-99/13:24 52407-16 Cold-1430 Davenport/Mr. Coogan 08-05-99/13:50 52407-17 Hot-1430 Davenport/Mr. Coogan 08-05-99/13:46 52407-18 Outside-1430 Davenport/Mr. Coogan 08-05-99/14:12 PARAMETER 52407-14 52407-15 52407-16 52407-17 52407-18 Copper (200.7) 0.062 <0.020 16 10 0.11 Copper, mg/l 08.12.99 08.12.99 08.12.99 08.12.99 08.12.99 Prep Date 08.16.99 08.16.99 08.16.99 08.16.99 08.16.99 Analysis Date 220 Hardness as CaCO3, mg/l 240 <3.3 <3.3 <3.3 Sulfate as SO4, mg/l <5.0 <5.0 7.9 12 <5.0 Analysis Date 08.09.99 08.09.99 08.09.99 08.09.99 08.09.99 15 15 Color, PCU 5 10 10 Analysis Date 08.05.99 08.05.99 08.05.99 08.05.99 08.05.99 Odor, TON <1 <1 <1 <1 <1 Analysis Date 08.05.99 08.05.99 08.05.99 08.05.99 08.05.99 7.7 7.6 7.7 7.6 pH, units 7.6 Analysis Date 08.05.99 08.05.99 08.05.99 08.05.99 08.05.99 Sulfide (376.2), mg/1 <0.10 <0.10 <0.10 <0.10 <0.10 08.11.99 08.11.99 08.11.99 08.11.99 08.11.99 Prep Date Analysis Date 08.11.99 08.11.99 08.11.99 08.11.99 08.11.99

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SAVANNAH LABORATORIES

& ENVIRONMENTAL SERVICES, INC.

LOG NO: B9-52407 Received: 05 AUG 99 Reported: 18 AUG 99 nan ist i die een gebrete gebrete in die een de de werde de werde de werde de bezeiten de de werde de beste de here de beste de beste de beste de here de de beste de Florida Legislature-Office Of Public Counsel 2308 Clara Kee Blvd Tallahassee, FL 32303 Project: Aloha Utilities Sampled By: Savannah Laboratories Code: 130190824 REPORT OF RESULTS Page 5 DATE/ LOG NO SAMPLE DESCRIPTION , QC REPORT FOR LIQUID SAMPLES TIME SAMPLED _____ 52407-11 Method Blank 52407-12 Accuracy (%Rec) 52407-13 Precision (%RPD) _____ PARAMETER 52407-11 52407-12 52407-13 Copper (200.7) 106 % Copper, mg/l <0.020 3.4 % 08.12.99 08.12.99 Prep Date _ _ _ 08.16.99 08.16.99 Analysis Date - - -Hardness as CaCO3, mg/l <3.3 113 % 1.6 % 95 % Sulfate as SO4, mg/l <5.0 5.8 % Analysis Date 08.09.99 08.09.99 ---Color, PCU <5 100 % 0 % 08.05.99 08.05.99 Analysis Date ---Odor, TON - - -<1 ---Analysis Date 08.05.99 - - ---pH, units 104 % 6.3 0.17 % Analysis Date 08.05.99 08.05.99 ----Sulfide (376.2), mg/l <0.10 104 % 0 % 08.11.99 Prep Date 08.11.99 ----Analysis Date 08.11.99 08.11.99 -----Method: EPA 600/4-79-020

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DOH Certification #'s: E84282, 84385, 87279, E87052

Kathy Sheffield Veroject Manager

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EXHIBIT TLB-5: TEST RESULTS OF FURTHER LAB TESTS OBTAINED



En & ENVIRONMENTAL SERVICES, INC.

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Case Narrative: SL Project B9-52407

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Date:	August 30, 1999
Client:	Florida Legislature - Office of Public Counsel
Project:	Aloba Utilities
Laboratory:	Savannah Laboratories and Environmental Services, Inc. Tampa Bay Division

Eighteen aqueous samples were received on August 5, 1999 for Copper, Hardness, Sulfate, Color, Odor, pH, and Sulfate analyses Sulfate was not detected in any of the samples. Copper was detected in varying concentrations, with the highest level in the sample identified as B952407-16 "Cold-1430 Davenport/Mr. Coogan". The client requested additional information about that sample. A qualitative anion screen was run, with a chloride result of ~30mg/1 and a sulfate result of ~20mg/1. These results as considered qualitative only.

The sample appeared to have a fine black precipitant in the bottom of the container. The appearance of the precipitant did not resemble copper sulfide, which would be more granular and would tend to suspend better and longer in solution. The precipitant in the sample bottle settled very quickly into the bottom of the container. It was more indicative of charcoal (carbon) fines in both appearance and settlability.

No significant QA/QC problems were encountered.

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Case Narrative: SL Project B9-52379

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Date:	August 30, 1999
Client:	Florida Legislature - Office of Public Counsel
Project:	Aloha Utilities
Laboratory:	Savannah Laboratories and Environmental Services, Inc. Tampa Bay Division

Ten aqueous samples of raw and finished water were received on August 4, 1999 for Copper, Hardness, Sulfate, Color, Odor, pH, and Sulfide analyses. Sulfate was not detected in any of the samples. Strong chlorine odors were detected in the finished water samples from MW-1, MW-3, and MW-9. The sample identified as "MW-9 (FINISH)" was analyzed on 8/25/99 for residual chlorine at client request. The result was 1.4 mg/l residual chlorine, even though the sample was analyzed three weeks outside holding time and had not been refrigerated, as per the analytical method. This would be indicative of a high residual chloride value at the time of sample collection.

No significant QA/QC problems were encountered with the original analyses.

SL SAVANNAH LABORATORIES & ENVIRONMENTAL SERVICES, INC. ANALYSIS REQUEST AND CHAIN OF CUSTODY RECORD	 ☐ 5102 LaRoche Avenue, Savannah, GA 31404 ☐ 2846 Industrial Plaza Drive, Tallahassee, FL 32301 ☐ 414 SW 12th Avenue, Deerfield Beach, FL 33442 ☐ 900 Lakeside Drive, Mobile, AL 36693 ☐ 6712 Benjamin Road, Suite 100, Tampa, FL 33634 ☐ 100 Alpha Drive, Suite 110, Destrehan, LA 70047
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